

Programme Outline

Day 1	Day 2
 Theory: Overview of laboratory safety Introduction to Workplace Safety and Health Act Common laboratory hazards Personal protection and safety facilities Safety Data Sheet (SDS) of chemicals Handling and disposal of chemical and physical wastes Management of hazardous chemicals Theory: Perform the basic calculations necessary for sample and solution preparation Basic concepts of mass, volume, concentration, number of moles, dilution and serial dilution Common laboratory apparatus such as glassware, weighing and measurement devices, mixing and heating equipment Calculate volume, mass, number of moles and concentration of samples Practical (Laboratory): Preparation of standard solutions/ serial dilution Prepare solutions from solid chemical or concentrated stock solutions Prepare standard solutions and perform serial dilution Check the absorbance value of standard solutions using UV-Vis spectrophotometry Plot and interpret the standard calibration graph 	Practical (Laboratory): Acid-base titration Fundamental concepts of titration Determine the end-point of the reaction using manual titration Calculate concentration of unknown solutions based on the volume of titrant Identify necessary precautions involved in a titration Theory: Concepts of laboratory quality control Good laboratory practices, i.e. documentation, standard operation procedures (SOP), equipment calibration, reagent preparation and labelling Reliability and accuracy in measurements Introduction of SPC charts Plot and analyse statistical process control (SPC)